

**REMARKS**

Claims 1-22 stand rejected under 35 U.S.C. § 103(a) as being unpatentable over U.S.

Patent No. 6,106,790 to Hsuing et al.

Hsuing et al. was cited as teaching a cleaning gas including  $\text{NF}_3$  (column 4, lines 9-13). The Examiner reads Hsuing et al. as also teaching that the cleaning gas may contain  $\text{SF}_6$  and/or  $\text{F}_2$  and also includes  $\text{N}_2$ ,  $\text{O}_2$ ,  $\text{CF}_4$ ,  $\text{C}_2\text{F}_6$ , and  $\text{CHF}_3$  (citing column 2, lines 61-67) with a 1%  $\text{NF}_3$  concentration (column 5, lines 21-22). Although acknowledging that Hsuing et al. fails to teach a cleaning gas containing  $\text{SF}_6$  in an amount of about 0.4-4.5 volume % and an inert gas in an amount of from 0.01 to 500 in terms of volume ratio, the Examiner considered Hsuing et al. as teaching the grouping of  $\text{CHF}_3$  and  $\text{SF}_6$  as fluorocarbons used for similar purposes in a semiconductor lab (citing column 4, lines 12-14) and further teaching a feed (gas) comprising 3 weight %  $\text{CHF}_3$  (citing column 5, lines 55-56).

The reason for rejection was that it would have been obvious to provide (a cleaning gas) containing 3 weight %  $\text{SF}_6$  because both ( $\text{SF}_6$  and  $\text{CHF}_3$ ) are used for similar purposes (citing column 4, lines 12-14).

Applicants traverse, and respectfully request the Examiner to reconsider for the following reasons.

**The Invention:**

As claimed in present claim 1, the invention is directed to a cleaning gas for removing deposits in semiconductor production equipment. The cleaning gas comprises an inert gas and at least two gases selected from  $\text{SF}_6$ ,  $\text{F}_2$  and  $\text{NF}_3$  excluding the combination of  $\text{F}_2$  and  $\text{NF}_3$  alone. The cleaning gas contains  $\text{SF}_6$  in an amount of about 0.4-4.5 vol %. Furthermore,  $\text{F}_2$  and/or  $\text{NF}_3$

is from 0.01 to 5 and the inert gas is from 0:01 to 500 in terms of the volume ratio assuming that  $\text{SF}_6$  is 1.

Hsuing et al.:

Turning to the cited prior art, Hsuing et al. surely teaches a cleaning gas containing  $\text{NF}_3$  for use in cleaning CVD reactor chambers (column 4, lines 9-12). The Examiner then goes on to assert that Hsuing et al. teaches that the cleaning gas may also contain  $\text{SF}_6$  and/or  $\text{F}_2$  and also a plurality of other gases (citing column 2, lines 61-62) including a 1 %  $\text{NF}_3$  concentration (citing column 5, lines 21-22). However, it is respectfully submitted that the Examiner mischaracterizes the reference and reads passages out of context.

Particularly, the passage at column 2, lines 61-67 describes a typical waste gas containing  $\text{NF}_3$  and a host of other components ordinarily collected and combined from the various unit operations in a semiconductor fabrication facility. Note, for example, the passage at column 4, lines 13-16 which describes that the FC consumption in the manufacturing processes (plural) is usually not complete, such that the exhaust stream will contain a mixture of FCs. Nowhere does Hsuing et al. teach, suggest, describe or disclose a cleaning gas as a single unit operation containing the large number of components described at column 2, lines 61-66, and in fact such cleaning gas does not exist. Rather, after collecting the exhaust from many unit operations using different gases, the object of Hsuing et al. is to selectively destroy  $\text{NF}_3$  so as to allow for recovery of other components. Note also that the passage at column 4, lines 12-13 describes that other FCs, such as  $\text{CF}_4$ ,  $\text{C}_2\text{F}_6$ , etc., are used in the semiconductor fabs for similar purposes. This

passage does not say that these other FCs are used in combination with  $\text{NF}_3$ , but rather teaches to the contrary.

Example 1 of Hsuing et al. cited by the Examiner describes treatment of a feed containing 1 %  $\text{NF}_3$  and nitrogen with a fluidized iron powder bed so as to evaluate  $\text{NF}_3$  conversion. This has nothing to do with a cleaning gas, and also has nothing to do with an exhaust gas other than possibly suggesting that a typical exhaust gas might contain 1 %  $\text{NF}_3$ .

No Motivation to Substitute  $\text{SF}_6$  for  $\text{CHF}_3$ :

The Examiner cites Hsuing et al. as teaching that both  $\text{CHF}_3$  and  $\text{SF}_6$  are fluorocarbons that can be used for similar purposes in a semiconductor lab, citing column 4, lines 12-14, and further cites Hsuing et al. as teaching a feed comprising 3 weight %  $\text{CHF}_3$ , citing column 5, lines 55-56. The Examiner then goes to conclude that it would have been obvious to substitute  $\text{SF}_6$  in the feed containing 3 weight %  $\text{CHF}_3$  because both  $\text{CHF}_3$  and  $\text{SF}_6$  are used for similar purposes.

Applicants respectfully disagree. The feed of Example 3 containing 3 weight %  $\text{CHF}_3$  is not an etching gas or a cleaning gas, but rather attempts to simulate a waste stream. As described bridging columns 5-6, because  $\text{CHF}_3$  is the most reactive of the four FCs used in the electronics industry,  $\text{CHF}_3$  was added to the test waste stream (called a "feedstock" in Example 3) to confirm that  $\text{NF}_3$  was selectively destroyed without converting  $\text{CHF}_3$ . As described at column 5, line 62, "no  $\text{CHF}_3$  conversion was observed".

Perhaps the Examiner considered that claim 1 reads on Example 3 of Hsuing et al. when  $\text{SF}_6$  is substituted for  $\text{CHF}_3$  in the waste gas. Although Example 3 employed the same type of iron powder loaded in the same reactor as described in Example 1, nowhere does Example 3

disclose that the same feed gas of Example 1 containing  $\text{NF}_3$  was used. To the contrary, Example 3 describes that the challenging feed stock "contained a nominal 3 weight %  $\text{CHF}_3$ " and there is no mention of  $\text{NF}_3$ . Aside from the above, Example 3 was a test to determine whether the most reactive of FCs would be converted in the reactor. This was not a test of etching rate or any other unit operation in a semiconductor fabrication, and thus there is no motivation or reason to even consider substituting  $\text{SF}_6$  for the  $\text{CHF}_3$ . Hsuing et al. wanted to know whether the most reactive of FCs would be converted in the reactor, and was not interested in less reactive  $\text{SF}_6$ .

Nothing to Optimize:

Lastly, the Examiner considered that it would have been obvious to optimize FCs because "discovering an optimum value or a result effective variable involves only routine skill in the art". However, there is simply nothing to optimize in Hsuing et al. The idea in Hsuing et al. is to convert  $\text{NF}_3$  without converting FCs. See column 4, lines 7-9 (a further objective of this invention is to selectively destroy  $\text{NF}_3$  in the presence of other fluorinated compounds (FCs)) and the disclosure at column 4, lines 18-19 (the semiconductor industry prefers the option of reclaiming and recycling the FCs for reuse). Regardless of the amount of FCs in the waste stream, the object of Hsuing et al. is to selectively destroy the  $\text{NF}_3$  without converting FCs. The exhaust gas can contain as much or as little FCs as is produced by the various unit operations in the semiconductor facility.

In summary, Hsuing et al. teaches a process for destroying  $\text{NF}_3$  in a gas comprising  $\text{NF}_3$ ,  $\text{N}_2$ ,  $\text{O}_2$ ,  $\text{F}_2$ ,  $\text{CF}_4$ ,  $\text{CHF}_3$ ,  $\text{SF}_6$  and  $\text{C}_2\text{F}_6$  by contacting the gas with metal particles capable of reacting with  $\text{NF}_3$ . The gas described therein is a typical waste gas containing  $\text{NF}_3$  and other

components ordinarily collected in a semiconductor fabrication facility, and is NOT a cleaning gas.

That is, Hsuing et al. discloses a waste gas, namely, a gas to be treated, while the mixed gas of the present invention is a cleaning gas. Not only does Hsuing et al. fail to describe the claimed cleaning gas, because it concerns a waste gas, there is nothing in Hsuing et al. which would suggest the desirability of modifying waste gas collected in a semiconductor fabrication facility to arrive at the claimed cleaning gas.

For the above reasons, it is respectfully submitted that the present claims are patentable over Hsuing et al., and withdrawal of the foregoing rejection under 35 U.S.C. § 103(a) is respectfully requested.

Withdrawal of all rejections and allowance of claims 1-22 is earnestly solicited.

In the event that the Examiner believes that it may be helpful to advance the prosecution of this application, the Examiner is invited to contact the undersigned at the local Washington, D.C. telephone number indicated below.

RESPONSE UNDER 37 C.F.R. § 1.111  
U.S. Application No. 10/088,306

Q60716

The USPTO is directed and authorized to charge all required fees, except for the Issue Fee and the Publication Fee, to Deposit Account No. 19-4880. Please also credit any overpayments to said Deposit Account.

Respectfully submitted,



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